

Mahumbane Consulting

Water Treatment Technical Data Sheet Library and Safety Data Sheet Format



This document is a structured resource pack for Mahumbane Consulting's website and technical handbook. It brings together ready-to-use **Technical Data Sheets (TDS)** for core water and wastewater treatment chemicals, together with a practical **Safety Data Sheet (SDS)** format and sample completed SDS content for key products.

Recommended website use

- Publish each TDS as an individual downloadable product sheet.
- Use the SDS format as the standard structure for supplier-specific safety documents.
- Review every sheet against the exact supplied grade, concentration, packaging, and transport classification before release.
- Pair this library with application guides for cooling towers, boilers, clarification, and disinfection.

Contents

| Section | Summary |
|---------|---|
| Part A | Technical Data Sheets for 20 common water and wastewater treatment chemicals and treatment products |
| Part B | Universal 16-section SDS format for Mahumbane-issued safety documentation |
| Part C | Completed sample SDS content for six high-priority products |

Important release note: this library is designed as a professional starting point. Values such as product concentration, grade, UN number, and transport class must be matched to the exact product supplied.

Part A - Technical Data Sheets

The following TDS pages are structured for direct use in Mahumbane product literature. Each sheet summarizes the product role, common applications, key handling considerations, and standard purchasing information.

Sodium Hypochlorite Solution

| Field | Details |
|---------------|--------------------------------|
| Product type | Disinfectant / Oxidizing Agent |
| Chemical name | Sodium Hypochlorite |
| Formula | NaOCl |
| CAS number | 7681-52-9 |
| UN number | UN1791 |

Description

A chlorine-based liquid oxidizer used for water disinfection, microbial control, odour control, and general sanitation across municipal and industrial systems.

Typical applications

- Potable water disinfection
- Wastewater disinfection
- Cooling tower biological control
- Odour control
- General sanitation

Typical properties

| Property | Value |
|--------------------|------------------------------|
| Appearance | Clear to yellow-green liquid |
| Available chlorine | Commonly 10-15% |
| pH | Typically 11-13 |
| Solubility | Completely miscible in water |
| Odour | Chlorine-like |

Recommended dosage range

Dose according to chlorine demand, retention time, and target free chlorine residual. Confirm by plant testing.

Handling and storage

Store cool, shaded, and ventilated. Keep away from acids, ammonia, reducing agents, and organic contamination. Use HDPE, PVC, or FRP compatible equipment.

PPE requirements

Chemical-resistant gloves, goggles or face shield, protective clothing, and respiratory protection where vapours may accumulate.

Packaging

25 L, 210 L drum, 1000 L IBC, or bulk tanker supply.

Important notes

Do not mix with acids. Toxic chlorine gas may be released.

Calcium Hypochlorite

| Field | Details |
|---------------|-------------------------|
| Product type | Disinfectant / Oxidizer |
| Chemical name | Calcium Hypochlorite |
| Formula | Ca(OCl)2 |
| CAS number | 7778-54-3 |
| UN number | UN1748 |

Description

A solid chlorinating compound used where stable solid disinfectant supply is preferred for municipal, industrial, and emergency treatment applications.

Typical applications

- Drinking water disinfection
- Emergency water treatment
- Swimming pool chlorination
- Wastewater disinfection

Typical properties

| Property | Value |
|--------------------|---|
| Appearance | White to off-white granules, powder, or tablets |
| Available chlorine | Often 65-70% |
| Odour | Chlorine-like |
| Solubility | Partially soluble in water |

Recommended dosage range

Apply according to required free chlorine residual and raw water quality.

Handling and storage

Store in a cool, dry area in tightly closed containers. Keep away from organic material, acids, oils, and reducing agents.

PPE requirements

Gloves, goggles, protective clothing, and dust mask or respirator if dust is generated.

Packaging

5 kg and 25 kg pails or drums.

Important notes

Strong oxidizer. Keep away from combustibles.

Chlorine Gas

| Field | Details |
|---------------|-------------------------|
| Product type | Disinfectant / Oxidizer |
| Chemical name | Chlorine |
| Formula | Cl ₂ |
| CAS number | 7782-50-5 |
| UN number | UN1017 |

Description

A highly effective gaseous disinfectant used in large municipal and industrial systems requiring controlled chlorination.

Typical applications

- Potable water treatment
- Wastewater disinfection
- Biological slime control
- Oxidation of contaminants

Typical properties

| Property | Value |
|-------------|-------------------------------|
| Appearance | Greenish-yellow gas |
| Odour | Pungent chlorine odour |
| Solubility | Moderately soluble in water |
| Corrosivity | Corrosive in moist conditions |

Recommended dosage range

Establish feed rate from system demand, contact time, and residual target. Use approved gas dosing equipment only.

Handling and storage

Store secured cylinders upright in a cool, dry, well-ventilated area with leak detection and emergency response controls.

PPE requirements

Chemical goggles, face shield, gloves, and chlorine-rated respiratory protection or SCBA for emergencies.

Packaging

Cylinders and ton containers.

Important notes

Highly toxic gas. Only trained personnel should handle it.

Ferric Chloride Solution

| Field | Details |
|---------------|-------------------|
| Product type | Coagulant |
| Chemical name | Ferric Chloride |
| Formula | FeCl ₃ |
| CAS number | 7705-08-0 |
| UN number | UN2582 |

Description

An acidic iron salt coagulant used to destabilize suspended matter, improve clarification, reduce phosphorus, and condition sludge.

Typical applications

- Wastewater clarification
- Industrial effluent treatment
- Phosphate removal
- Sludge dewatering
- Sewer odour control

Typical properties

| Property | Value |
|------------------|-----------------------------|
| Appearance | Dark brown liquid |
| Nature | Strongly acidic |
| Solubility | Completely soluble in water |
| Equipment impact | Corrosive to many metals |

Recommended dosage range

Optimize by jar testing or site trials to match solids load, alkalinity, and phosphorus reduction target.

Handling and storage

Use acid-resistant tanks and lines such as PVC, HDPE, rubber-lined steel, or FRP. Avoid mild steel and aluminium.

PPE requirements

Acid-resistant gloves, goggles, face shield, apron or chemical suit, and rubber boots.

Packaging

25 L, 200 L, 1000 L IBC, or bulk supply.

Important notes

Corrosive and staining. Good housekeeping is essential.

Ferric Sulphate

| Field | Details |
|---------------|---|
| Product type | Coagulant |
| Chemical name | Ferric Sulphate |
| Formula | Fe ₂ (SO ₄) ₃ |
| CAS number | 10028-22-5 |
| UN number | Verify supplied grade |

Description

An iron-based coagulant used for clarification, colour reduction, and phosphorus removal in water and wastewater systems.

Typical applications

- Clarification
- Phosphorus removal
- Sludge conditioning

Typical properties

| Property | Value |
|------------|---|
| Appearance | Brown or yellow liquid / solid depending on grade |
| Nature | Acidic solution |
| Solubility | Soluble in water |

Recommended dosage range

Establish by testing against plant pH, alkalinity, and treatment target.

Handling and storage

Store in compatible acid-resistant containers. Protect transfer areas from corrosion.

PPE requirements

Gloves, goggles, face shield, protective clothing.

Packaging

Drums, IBCs, bulk, or bags depending on grade.

Important notes

Use actual supplied specification for transport classification.

Aluminium Sulphate

| Field | Details |
|---------------|---|
| Product type | Coagulant |
| Chemical name | Aluminium Sulphate |
| Formula | Al ₂ (SO ₄) ₃ |
| CAS number | 10043-01-3 |
| UN number | Verify supplied grade |

Description

A conventional coagulant widely used for turbidity reduction and clarification in potable and industrial water treatment.

Typical applications

- Potable water clarification
- Wastewater treatment
- Paper and pulp processes
- Sludge conditioning

Typical properties

| Property | Value |
|--------------------|---|
| Appearance | White to grey solid or clear to pale liquid |
| Solubility | Soluble in water |
| Solution character | Acidic in solution |

Recommended dosage range

Determine by jar testing; performance depends on raw water pH and alkalinity.

Handling and storage

Store dry if solid. Protect liquid product from contamination and use compatible equipment.

PPE requirements

Gloves, goggles, protective clothing, and dust protection for powder grades.

Packaging

25 kg bags, drums, or bulk liquid.

Important notes

Alkalinity demand should be considered when dosing.

Polyaluminium Chloride

| Field | Details |
|---------------|-----------------------------------|
| Product type | Coagulant |
| Chemical name | Polyaluminium Chloride |
| Formula | Variable polymeric aluminium salt |
| CAS number | 1327-41-9 |
| UN number | Verify supplied grade |

Description

A pre-hydrolyzed aluminium coagulant that often achieves good clarification over a wider pH range with lower sludge generation than alum.

Typical applications

- Potable water treatment
- Industrial wastewater treatment
- Turbidity and colour reduction
- Pretreatment before filtration

Typical properties

| Property | Value |
|-----------------|---------------------------------|
| Appearance | Yellow liquid or powder |
| Solubility | Soluble in water |
| Operating range | Effective over a broad pH range |

Recommended dosage range

Confirm with jar testing and plant trials.

Handling and storage

Store in compatible plastic or lined tanks; protect powder from moisture.

PPE requirements

Gloves, goggles, protective clothing.

Packaging

Bags, drums, IBCs.

Important notes

Useful where reduced sludge and robust performance are required.

Polyacrylamide Flocculant

| Field | Details |
|---------------|---|
| Product type | Flocculant |
| Chemical name | Polyacrylamide |
| Formula | (C ₃ H ₅ NO) _n |
| CAS number | 9003-05-8 |
| UN number | Usually not regulated |

Description

A high molecular weight polymer supplied as anionic, cationic, or nonionic grades to agglomerate suspended particles and improve settling and dewatering.

Typical applications

- Wastewater clarification
- Sludge thickening
- Sludge dewatering
- Mineral processing
- Raw water clarification

Typical properties

| Property | Value |
|----------------|-----------------------------------|
| Appearance | White granular powder or emulsion |
| Solubility | Water soluble after make-up |
| Special hazard | Severe slip hazard when wet |

Recommended dosage range

Select grade and dose through bench testing. Make-up concentration and aging time affect performance.

Handling and storage

Keep dry and sealed. Protect from moisture. Clean spills immediately to prevent slips.

PPE requirements

Gloves, goggles, dust mask when handling powder, protective clothing.

Packaging

25 kg bags or drums for emulsion products.

Important notes

Correct polymer choice is critical to treatment performance.

Caustic Soda

| Field | Details |
|---------------|------------------------------|
| Product type | Alkali / pH Adjuster |
| Chemical name | Sodium Hydroxide |
| Formula | NaOH |
| CAS number | 1310-73-2 |
| UN number | UN1824 solid / UN1823 liquid |

Description

A strong alkali used to increase pH, neutralize acidic streams, support alkalinity, and assist with industrial cleaning applications.

Typical applications

- pH increase
- Neutralization of acidic streams
- CIP cleaning
- Regeneration systems

Typical properties

| Property | Value |
|--------------------|---------------------------------------|
| Appearance | White flakes / pearls or clear liquid |
| Character | Strongly alkaline |
| Solubility | Highly soluble in water |
| Dilution behaviour | Heat generated when diluted |

Recommended dosage range

Determine from acidity, alkalinity, and treatment target.

Handling and storage

Store in corrosion-resistant containers. Keep tightly closed. Add product to water carefully during dilution.

PPE requirements

Alkali-resistant gloves, goggles, face shield, protective clothing, and boots.

Packaging

25 kg bags, drums, IBCs, or bulk.

Important notes

Never add water rapidly into concentrated caustic.

Lime

| Field | Details |
|---------------|------------------------------------|
| Product type | Alkali / Neutralizing Agent |
| Chemical name | Calcium Oxide or Calcium Hydroxide |
| Formula | CaO / Ca(OH) ₂ |
| CAS number | 1305-78-8 / 1305-62-0 |
| UN number | Verify supplied grade |

Description

A versatile alkaline reagent used for pH adjustment, softening, metal precipitation, sludge stabilization, and neutralization.

Typical applications

- Neutralization
- Metal precipitation
- Sludge treatment
- Water softening

Typical properties

| Property | Value |
|----------------|-------------------|
| Appearance | White powder |
| Character | Alkaline |
| Water reaction | Reacts with water |

Recommended dosage range

Determine according to pH target and treatment chemistry.

Handling and storage

Store dry and protect from moisture ingress. Control dust during handling.

PPE requirements

Gloves, goggles, dust mask, protective clothing.

Packaging

25 kg bags, bulk bags, or bulk tanker for powder systems.

Important notes

Dust may irritate the eyes and respiratory tract.

Hydrochloric Acid

| Field | Details |
|---------------|-------------------|
| Product type | Acid / pH Reducer |
| Chemical name | Hydrochloric Acid |
| Formula | HCl |
| CAS number | 7647-01-0 |
| UN number | UN1789 |

Description

A strong mineral acid used for pH reduction, descaling, resin regeneration, and industrial cleaning.

Typical applications

- pH reduction
- Resin regeneration
- Descaling
- Industrial cleaning

Typical properties

| Property | Value |
|------------|---------------------------------|
| Appearance | Clear to slightly yellow liquid |
| Odour | Strong acidic odour |
| Character | Corrosive |

Recommended dosage range

Determine from alkalinity and pH reduction requirement.

Handling and storage

Store in acid-resistant materials with ventilation. Keep away from alkalis, oxidizers, and reactive metals.

PPE requirements

Acid-resistant gloves, goggles, face shield, chemical suit or apron.

Packaging

Drums, IBCs, or bulk.

Important notes

Reacts with metals to release hydrogen gas.

Sulphuric Acid

| Field | Details |
|---------------|--------------------------------|
| Product type | Acid / pH Reducer |
| Chemical name | Sulphuric Acid |
| Formula | H ₂ SO ₄ |
| CAS number | 7664-93-9 |
| UN number | UN1830 |

Description

A strong acid used for pH control, ion exchange regeneration, and industrial process treatment.

Typical applications

- pH control
- Ion exchange regeneration
- Industrial cleaning
- Process adjustment

Typical properties

| Property | Value |
|--------------------|------------------------------|
| Appearance | Clear oily liquid |
| Character | Strong acid |
| Dilution behaviour | Highly exothermic with water |

Recommended dosage range

Establish according to plant requirement and alkalinity.

Handling and storage

Store in approved acid systems. Always add acid to water during dilution.

PPE requirements

Acid-resistant gloves, goggles, face shield, chemical-resistant suit.

Packaging

Drums, IBCs, or bulk tanker.

Important notes

Severe burns possible. Strong heat is generated on dilution.

Sodium Bisulphite

| Field | Details |
|---------------|---|
| Product type | Reducing Agent / Dechlorination Chemical |
| Chemical name | Sodium Bisulphite |
| Formula | NaHSO ₃ |
| CAS number | 7631-90-5 |
| UN number | Verify concentration and transport status |

Description

A reducing agent used to remove residual chlorine and other oxidants from treated water and process streams.

Typical applications

- Dechlorination
- Reducing residual oxidants
- Oxygen scavenging in selected applications

Typical properties

| Property | Value |
|------------|---|
| Appearance | Clear liquid or powder depending on grade |
| Odour | Sulphurous odour |
| Character | Reducing agent |

Recommended dosage range

Set according to oxidant residual and target final concentration.

Handling and storage

Protect from air oxidation. Store in a cool area and keep sealed.

PPE requirements

Gloves, goggles, protective clothing.

Packaging

Drums, IBCs, or bags.

Important notes

May release sulphur dioxide under acidic conditions.

Sodium Metabisulphite

| Field | Details |
|---------------|---|
| Product type | Reducing Agent |
| Chemical name | Sodium Metabisulphite |
| Formula | Na ₂ S ₂ O ₅ |
| CAS number | 7681-57-4 |
| UN number | Verify supplied grade |

Description

A solid reducing agent commonly used for dechlorination and membrane pretreatment where oxidants must be removed.

Typical applications

- Dechlorination
- Membrane pretreatment
- Reducing residual oxidants

Typical properties

| Property | Value |
|------------|--------------------------|
| Appearance | White crystalline powder |
| Odour | Sulphurous odour |
| Solubility | Soluble in water |

Recommended dosage range

Apply according to residual oxidant concentration and process target.

Handling and storage

Keep dry and sealed. Avoid acid contamination.

PPE requirements

Gloves, goggles, dust mask.

Packaging

25 kg bags.

Important notes

May liberate sulphur dioxide in acidic conditions.

Hydrogen Peroxide

| Field | Details |
|---------------|-------------------------------|
| Product type | Oxidizer |
| Chemical name | Hydrogen Peroxide |
| Formula | H ₂ O ₂ |
| CAS number | 7722-84-1 |
| UN number | Depends on concentration |

Description

An oxidizing agent used for odour control, sulphide oxidation, and specialized treatment processes.

Typical applications

- Oxidation of sulphides
- Odour control
- Advanced oxidation systems
- Industrial effluent treatment

Typical properties

| Property | Value |
|------------|--|
| Appearance | Clear liquid |
| Character | Oxidizing agent |
| Behaviour | Decomposes in contact with contaminants or catalysts |

Recommended dosage range

Set by oxidation demand and treatment objective.

Handling and storage

Store in vented approved containers. Protect from contamination, heat, metals, and organics.

PPE requirements

Gloves, goggles or face shield, protective clothing.

Packaging

Drums or IBCs.

Important notes

Concentrated grades are hazardous strong oxidizers.

Activated Carbon

| Field | Details |
|---------------|-------------------------------------|
| Product type | Adsorbent |
| Chemical name | Activated Carbon |
| Formula | - |
| CAS number | 7440-44-0 |
| UN number | Usually not regulated; verify grade |

Description

A high surface area adsorbent used to remove organics, colour, odours, and residual disinfectants.

Typical applications

- Taste and odour control
- Organics removal
- Colour reduction
- Polishing treatment

Typical properties

| Property | Value |
|------------------|--------------------------|
| Appearance | Black granules or powder |
| Special property | High surface area |
| Solubility | Insoluble in water |

Recommended dosage range

Dose and contact time depend on contaminant load and process configuration.

Handling and storage

Keep dry and avoid dust generation. Store away from strong oxidizers.

PPE requirements

Gloves, dust mask, goggles.

Packaging

25 kg bags or bulk bags.

Important notes

Wet spent carbon may deplete oxygen in confined spaces.

Sodium Aluminate

| Field | Details |
|---------------|------------------------|
| Product type | Coagulant Aid / Alkali |
| Chemical name | Sodium Aluminate |
| Formula | NaAlO ₂ |
| CAS number | 1302-42-7 |
| UN number | Verify supplied grade |

Description

An alkaline aluminium reagent used as a coagulant aid, alkalinity source, and phosphorus removal aid.

Typical applications

- Coagulation aid
- Phosphorus removal
- Clarification

Typical properties

| Property | Value |
|------------|-----------------------|
| Appearance | White solid or liquid |
| Character | Strongly alkaline |

Recommended dosage range

Determine through testing against water chemistry and clarification objective.

Handling and storage

Store in alkali-resistant containers.

PPE requirements

Gloves, goggles, face shield, protective clothing.

Packaging

Bags, drums, or bulk.

Important notes

Use actual supplied spec for concentrations and shipping data.

Antiscalant

| Field | Details |
|---------------|-------------------|
| Product type | Scale Inhibitor |
| Chemical name | Proprietary blend |
| Formula | Mixture |
| CAS number | Mixture |
| UN number | Product specific |

Description

A proprietary scale inhibitor used to suppress precipitation of calcium, magnesium, silica, and related salts in membrane and recirculating systems.

Typical applications

- RO pretreatment
- Boiler feedwater systems
- Cooling water systems

Typical properties

| Property | Value |
|------------|---------------|
| Appearance | Clear liquid |
| Solubility | Water soluble |

Recommended dosage range

Use according to system recovery, water chemistry, and supplier-approved dosage model.

Handling and storage

Keep sealed and protect from freezing or extreme heat.

PPE requirements

Gloves and goggles.

Packaging

Drums, IBCs, or bulk.

Important notes

Use a membrane-compatible grade where relevant.

Corrosion Inhibitor

| Field | Details |
|---------------|----------------------------|
| Product type | Corrosion Control Chemical |
| Chemical name | Proprietary blend |
| Formula | Mixture |
| CAS number | Mixture |
| UN number | Product specific |

Description

A treatment blend used to protect metal surfaces in cooling towers, closed loops, and recirculating industrial systems.

Typical applications

- Cooling towers
- Closed chilled water loops
- Industrial recirculating systems

Typical properties

| Property | Value |
|---------------|--|
| Appearance | Typically liquid blend |
| Program basis | Composition depends on treatment program |

Recommended dosage range

Program-specific. Confirm by monitoring corrosion rates and water chemistry.

Handling and storage

Store in sealed containers and prevent contamination.

PPE requirements

Gloves and goggles as minimum protection.

Packaging

Drums, IBCs, or bulk.

Important notes

System metallurgy determines the preferred program.

Non-Oxidizing Biocide

| Field | Details |
|---------------|-------------------|
| Product type | Microbiocide |
| Chemical name | Proprietary blend |
| Formula | Mixture |
| CAS number | Mixture |
| UN number | Product specific |

Description

A microbiocide used to control bacteria, algae, and biofilm in cooling towers and process water loops.

Typical applications

- Cooling towers
- Process water loops
- Slime control

Typical properties

| Property | Value |
|---------------|---|
| Appearance | Usually clear to pale liquid |
| Program basis | Application depends on biological load and retention time |

Recommended dosage range

Apply shock or continuous dose according to microbial monitoring and system design.

Handling and storage

Store tightly closed in a cool, ventilated area away from incompatible chemicals.

PPE requirements

Gloves, goggles, and protective clothing as directed by the product SDS.

Packaging

Drums, IBCs, or bulk.

Important notes

Alternate biocides may be used to control resistance and biofilm.

Part B - Universal Safety Data Sheet Format

Use the structure below for each Mahumbane product SDS. The language must be completed with the exact supplied product concentration, ingredient disclosure, GHS classification, transport details, and emergency contact information.

| SDS section | What must be completed |
|---|--|
| 1. Identification | Product name, recommended use, supplier details, emergency contact number |
| 2. Hazard identification | GHS classification, signal word, hazard statements, precautionary statements |
| 3. Composition / information on ingredients | Chemical identity, CAS number, concentration, mixture disclosure |
| 4. First aid measures | Inhalation, skin, eye, and ingestion response; key symptoms |
| 5. Fire fighting measures | Suitable extinguishing media, special hazards, protective equipment |
| 6. Accidental release measures | Personal precautions, environmental precautions, cleanup method |
| 7. Handling and storage | Safe handling practices, storage conditions, incompatible materials |
| 8. Exposure controls / personal protection | Exposure limits, engineering controls, PPE |
| 9. Physical and chemical properties | Appearance, odour, pH, solubility, density, other relevant properties |
| 10. Stability and reactivity | Reactivity, stability, hazardous reactions, incompatible materials |
| 11. Toxicological information | Likely routes of exposure and acute/chronic effects |
| 12. Ecological information | Aquatic toxicity, persistence, degradability, mobility |
| 13. Disposal considerations | Waste handling and container disposal guidance |
| 14. Transport information | UN number, proper shipping name, class, packing group |
| 15. Regulatory information | Applicable legislation and chemical control references |
| 16. Other information | Issue date, revision date, disclaimer, preparer notes |

Mahumbane release recommendation

For legal and customer-facing use, each SDS should be reviewed against the actual manufacturer specification, transport classification, and South African handling requirements before issue.

Part C - Completed Sample SDS Content

These sample pages show how Mahumbane can present completed SDS content for frequently supplied products. They are suitable as editable working drafts and should be finalized against actual supplied product data.

Sodium Hypochlorite Solution

| Section | Content |
|---|---|
| 1. Identification | Recommended use: Water disinfection and oxidation. Supplier: Mahumbane Consulting (Pty) Ltd. |
| 2. Hazard identification | Corrosive, oxidizing, and hazardous to aquatic life. Causes severe skin burns and eye damage. Contact with |
| 3. Composition / information on ingredients | Sodium hypochlorite in aqueous solution, typically 10-15% available chlorine. |
| 4. First aid measures | Inhalation: move to fresh air. Skin: wash with plenty of water. Eyes: rinse for at least 15 minutes and obtain ur |
| 5. Fire fighting measures | Not flammable, but may intensify surrounding fire as an oxidizer. Use media suitable for the surrounding fire. |
| 6. Accidental release measures | Wear PPE, ventilate area, contain spill, and prevent concentrated release to drains or watercourses. |
| 7. Handling and storage | Store cool and out of direct sunlight. Keep away from acids, ammonia, reducing agents, and organics. |
| 8. Exposure controls / personal protection | Use goggles, face shield, gloves, apron, and suitable ventilation. Respiratory protection may be needed when |
| 9. Physical and chemical properties | Yellow-green liquid with chlorine odour and alkaline pH. |
| 10. Stability and reactivity | Decomposes with heat, light, and contamination. Reacts with acids to release chlorine gas. |
| 11. Toxicological information | Corrosive to skin, eyes, and mucous membranes. |
| 12. Ecological information | Very toxic to aquatic organisms. |
| 13. Disposal considerations | Dispose of product and contaminated containers according to local regulations. Do not discharge concentrate |
| 14. Transport information | UN1791 - Hypochlorite solution - Class 8 - Packing Group III. |
| 15. Regulatory information | Subject to applicable chemical safety and transport regulations. |
| 16. Other information | Issue actual supplier revision date before publication. |

Ferric Chloride Solution

| Section | Content |
|---|---|
| 1. Identification | Recommended use: Coagulation and phosphate removal. |
| 2. Hazard identification | Corrosive. Causes severe skin burns and eye damage. May be corrosive to metals. |
| 3. Composition / information on ingredients | Ferric chloride in aqueous solution. |
| 4. First aid measures | Fresh air for inhalation. Flush skin and eyes with plenty of water. Seek medical attention. |
| 5. Fire fighting measures | Non-flammable. Use extinguishing media suitable for the surrounding fire. |
| 6. Accidental release measures | Contain spill, absorb with inert material, and prevent discharge to drains. |
| 7. Handling and storage | Store in corrosion-resistant containers. Keep away from bases and reactive metals. |
| 8. Exposure controls / personal protection | Use corrosion-resistant gloves, goggles, face shield, apron, and boots. |
| 9. Physical and chemical properties | Dark brown acidic liquid. |
| 10. Stability and reactivity | Stable under normal conditions. Reacts with alkalis and some metals. |

| | |
|-------------------------------|--|
| 11. Toxicological information | Corrosive to tissue. Harmful if swallowed. |
| 12. Ecological information | Avoid uncontrolled release. |
| 13. Disposal considerations | Dispose of according to regulations. |
| 14. Transport information | UN2582 - Ferric chloride solution - Class 8 - Packing Group III. |
| 15. Regulatory information | Subject to applicable chemical safety and transport regulations. |
| 16. Other information | Complete with actual product concentration before issue. |

Caustic Soda / Sodium Hydroxide

| Section | Content |
|---|---|
| 1. Identification | Recommended use: pH adjustment and neutralization. |
| 2. Hazard identification | Corrosive. Causes severe skin burns and eye damage. |
| 3. Composition / information on ingredients | Sodium hydroxide solid or solution. |
| 4. First aid measures | Flush immediately with plenty of water. Seek urgent medical attention for eye exposure. |
| 5. Fire fighting measures | Non-flammable. |
| 6. Accidental release measures | Contain spill. Avoid contact. Neutralize only if trained and authorized. |
| 7. Handling and storage | Store dry and tightly closed. Keep away from acids and reactive metals. |
| 8. Exposure controls / personal protection | Use gloves, face shield, goggles, apron, and boots. |
| 9. Physical and chemical properties | White solid or clear liquid. Strong alkali. |
| 10. Stability and reactivity | Reacts with acids. Heat is generated on dilution. |
| 11. Toxicological information | Severely corrosive to skin and eyes. |
| 12. Ecological information | May significantly alter water pH if released. |
| 13. Disposal considerations | Dispose according to local regulations. |
| 14. Transport information | UN1823 liquid / UN1824 solid - Class 8. |
| 15. Regulatory information | Subject to applicable chemical safety and transport regulations. |
| 16. Other information | Confirm actual concentration and packaging before release. |

Hydrochloric Acid

| Section | Content |
|---|---|
| 1. Identification | Recommended use: pH reduction and cleaning. |
| 2. Hazard identification | Corrosive. Causes severe burns and eye damage. May cause respiratory irritation. Corrosive to metals. |
| 3. Composition / information on ingredients | Hydrogen chloride in water. |
| 4. First aid measures | Fresh air. Flush skin and eyes immediately with water for at least 15 minutes. Seek urgent medical attention. |
| 5. Fire fighting measures | Non-flammable. Use extinguishing media suitable for surrounding materials. |
| 6. Accidental release measures | Ventilate, contain, and prevent entry to drains. Use trained personnel only. |

| | |
|--|---|
| 7. Handling and storage | Store in acid-resistant containers with ventilation. Keep away from alkalis, oxidizers, cyanides, and metals. |
| 8. Exposure controls / personal protection | Acid-resistant gloves, goggles, face shield, protective clothing. |
| 9. Physical and chemical properties | Clear to slightly yellow acidic liquid with sharp odour. |
| 10. Stability and reactivity | Reacts with alkalis and many metals; hydrogen gas may be generated. |
| 11. Toxicological information | Corrosive to skin, eyes, and respiratory tissue. |
| 12. Ecological information | Avoid uncontrolled release. |
| 13. Disposal considerations | Dispose according to regulations. |
| 14. Transport information | UN1789 - Hydrochloric acid - Class 8. |
| 15. Regulatory information | Subject to applicable chemical safety and transport regulations. |
| 16. Other information | Confirm concentration before issue. |

Sulphuric Acid

| Section | Content |
|---|--|
| 1. Identification | Recommended use: pH control and regeneration. |
| 2. Hazard identification | Corrosive. Causes severe burns and eye damage. |
| 3. Composition / information on ingredients | Sulphuric acid in water or concentrated form. |
| 4. First aid measures | Flush skin and eyes immediately with water and seek urgent medical attention. |
| 5. Fire fighting measures | Non-flammable. Use media suitable for the surrounding fire. |
| 6. Accidental release measures | Contain spill and keep away from water addition unless trained response procedures are in place. |
| 7. Handling and storage | Store in approved acid systems. Always add acid to water during dilution. |
| 8. Exposure controls / personal protection | Acid-resistant gloves, goggles, face shield, and chemical-resistant suit. |
| 9. Physical and chemical properties | Clear oily acidic liquid. |
| 10. Stability and reactivity | Strongly exothermic with water and reactive with many materials. |
| 11. Toxicological information | Severely corrosive to tissue. |
| 12. Ecological information | Avoid release to the environment. |
| 13. Disposal considerations | Dispose according to regulations. |
| 14. Transport information | UN1830 - Sulphuric acid - Class 8. |
| 15. Regulatory information | Subject to applicable chemical safety and transport regulations. |
| 16. Other information | Confirm actual concentration and packaging before issue. |

Polyacrylamide Flocculant

| Section | Content |
|--------------------------|---|
| 1. Identification | Recommended use: Flocculation and solids separation. |
| 2. Hazard identification | Usually not highly hazardous, but dust may irritate eyes and lungs. Spills create severe slip hazard. |

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| 3. Composition / information on Polyacrylate | Polycrylate polymer supplied as powder or emulsion. |
| 4. First aid measures | Flush eyes with water. Wash skin. Move to fresh air if dust is inhaled. |
| 5. Fire fighting measures | Use extinguishing media suitable for surrounding materials. |
| 6. Accidental release measures | Avoid raising dust. Clean spills immediately because wet polymer is extremely slippery. |
| 7. Handling and storage | Keep dry. Avoid dust generation. Seal containers after use. |
| 8. Exposure controls / personal protection | Use gloves, goggles, and dust mask where needed. |
| 9. Physical and chemical properties | White powder or emulsion; slippery when wet. |
| 10. Stability and reactivity | Stable under normal storage conditions. |
| 11. Toxicological information | May irritate eyes, skin, or respiratory tract under dusty conditions. |
| 12. Ecological information | Avoid large uncontrolled releases. |
| 13. Disposal considerations | Dispose according to local regulations. |
| 14. Transport information | Usually not regulated for transport. |
| 15. Regulatory information | Subject to applicable workplace handling requirements. |
| 16. Other information | Confirm specific grade and ionic type before release. |

Appendix - Fast Website Resource Structure

| Website folder | Recommended contents |
|--------------------------|---|
| Technical Data Sheets | One PDF per product with the TDS format used in Part A |
| Safety Data Sheets | One product-specific SDS per chemical using the 16-section format |
| Application Guides | Cooling tower treatment, boiler treatment, clarification, disinfection, chemical storage |
| PPE Guides | Chemical handling PPE matrix, eye protection, gloves, respiratory protection |
| Compliance / disclaimers | Document control, revision date, temporary generic disclaimer, request-for-latest-copy note |

Prepared for Mahumbane Consulting website development and technical resource planning.